



# QUANTUM DOT CHEMICAL STABILITY AND THEIR INTERACTION WITH ZEBRAFISH EMBRYOS



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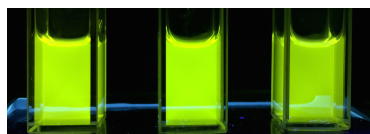
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## Background

Semiconductor quantum dots (QD) are nanoparticles with optical and electronic properties. The nanometric dimensions of QD result in unique photophysical properties, which include high brightness in the visible range of the electromagnetic spectrum, high photostability, narrow emission peaks, and size dependent emission tunability. The unique photophysical properties of luminescent semiconductor QDs enable their use in biological imaging, labeling and detection, and as optical elements in consumer electronics, light emitting displays, and photovoltaic devices. Semiconductor QDs have been drawing wide spread interest, however it is known that QDs that contain heavy metals, such as cadmium, could be toxic in high concentrations. There is concern that the wide scale use will have adverse impacts on human health and the environment.



## Materials and Methods

- Core Chemistry: **Cadmium selenide**
- Shell Chemistry: **Zinc sulfide**
- Surface Chemistry:
  - Pre-Cap Exchange: Triethylamine (TEA)
  - Triethylphosphine oxide (TOPO)
  - Post-Cap Exchange: Dihydrolipoic acid- polyethylene glycol-methoxy (**DHLA PEG-OCH<sub>3</sub>**)
- Colloidal stability of CdSe/ZnS/DHLA(750MW)PEG-OCH<sub>3</sub> was analyzed through UV-VIS absorbance spectroscopy, steady-state fluorescence spectroscopy and fluorescence lifetime spectroscopy measurements over the course of a week.

## Aim

To understand if the surface chemistry will alter the stability of luminescent quantum dots in biological solutions and if it can be used as a tool to regulate levels of nanotoxicity in zebrafish embryos.

## Results and Analysis

The QD that are capped with DHLA(750MW)PEG-OCH<sub>3</sub> ligand show colloidal stability in all buffer solutions with the exception of minimal media, and pH levels of 4 & 5.

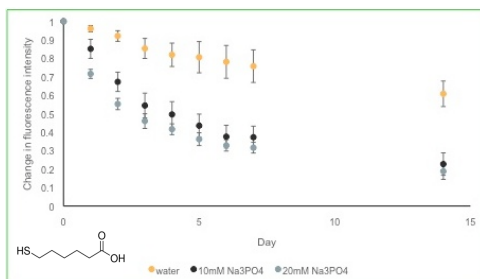


Figure 1. CdSe/ZnS/MUA QD in salt solutions over time

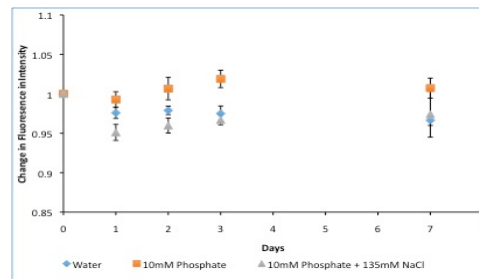


Figure 2. CdSe/ZnS/DHLA (750) PEG-OCH<sub>3</sub> QD in salt solutions over time

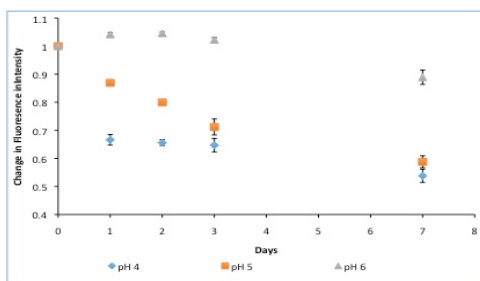


Figure 3. CdSe/ZnS/DHLA (750) PEG-OCH<sub>3</sub> QD in various pH levels over time

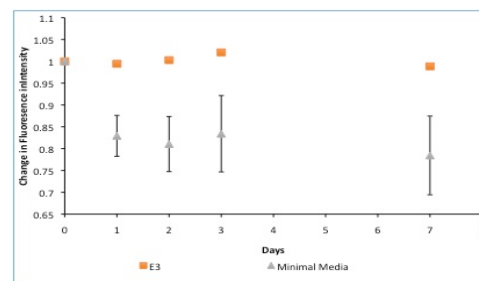


Figure 4. CdSe/ZnS/DHLA (750) PEG-OCH<sub>3</sub> QD in various media over time

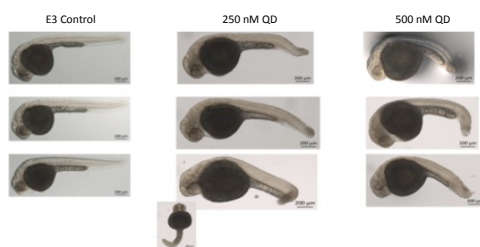


Figure 5. Brightfield Micrographs of 24hpf embryos exposed to CdSe/ZnS/DHLA (750) PEG-OCH<sub>3</sub>

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## Conclusions

- Altering the surface chemistry can affect the colloidal stability of QD in biological solutions.
- Cadmium based QD are still toxic regardless of surface chemistry.

**Impact statement: Through this research, we understand the specific chemical and physical interactions between Cadmium based Quantum Dots and living organisms.**

## Future Direction

- Continue stability studies with charged ligands
- Use alternatives to cadmium based cores

## References

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